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Exam. Code : 103205

Subject Code: 8053

B.A./B.Sc. 5th Semester (Old Sylb 2017)

CHEMISTRY (PHYSICAL CHEMISTRY-III)

Time Allowed—3 Hours]

[Maximum Marks—35

- Note:—(1) Part A is compulsory. Each question carries 1 mark.
 - (2) Attempt **TWO** questions each from the Sections **I**, **II** and **III** in Part **B**. Each question carries **4.5** marks.

PART-A

- 1. Define corrosion and what are its types?
- 2. Distinguish between electrolytic and galvanic cells.
- 3. Write Ostwald dilution law explaining each term.
- 4. What is nuclear reaction cross section?
- 5. Define nuclear spin and nuclear momentum.
- 6. What is Raman Spectrum?
- 7. What is qualitative relation between force constant and bond energies?
- 8. Write selection rules for electronic transitions and their significance. 1×8

PART—B SECTION—I

- 9. (a) Explain Kohlaursch law and what is its basis?
 - (b) What is Arrhenius theory of electrolytes and what are its limitations? 2.5.2.0
- 10. (a) Define single electrode potential.
 - (b) How can ΔG , ΔH and K can be determined using EMF? 1.0,3.5

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(Contd.)

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- 11. (a) Define pH and pKa.
 - (b) Describe concentration cells with and without transport with suitable examples. 1.0,3.5

SECTION—II

- 12. (a) Distinguish between nuclear fission and nuclear fusion with appropriate examples.
 - (b) Write applications of radioactivity. 2.0,2.5
- 13. (a) Describe Soddy Fajan group displacement law.
 - (b) Explain nuclear structure, mass defects and nuclear stability. 2.5,2.0
- 14. (a) What are the means of measuring nuclear decay?
 - (b) What are high energy nuclear reactions? 3.0,1.5

SECTION—III

- 15. (a) What is role of isotope effect on rotational spectrum?
 - (b) To a good approximation, the microwave spectrum of $\mathrm{H}^{35}\mathrm{Cl}$ consists of a series of equally spaced lines, separated by $6.26 \times 10^{11}\mathrm{Hz}$. Calculate bond length of $\mathrm{H}^{35}\mathrm{Cl}$.
- 16. (a) Explain P, Q and R branches in vibrational rotational spectrum.
 - (b) Draw energy levels of harmonic oscillator. How force constant can be evaluated? 2,2.5
- 17. (a) What is Frank Condon principle?
 - (b) What are selection rules for electronic transitions?
 - (c) Explain origin of Raman spectrum. 1.0,1.0,2.5